

# Car-Parrinello Molecular Dynamics

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Moral: "A man dreams of a miracle and wakes up with loaves of bread"

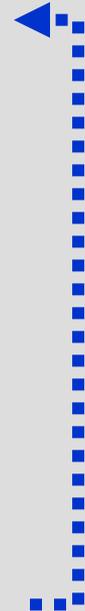
Erich Maria Remarque

# Molecular Dynamics Loop

(1) Compute Forces on atoms,  $F_i(t)$  for current atomic configuration,  $R_i(t)$

$F_i(t) \leftarrow$

- calculate using classical potentials  
(can do large systems and long simulation times)
- calculate directly from first principles by solving many-electron Schrödinger equations  
(can treat very complex chemistry, but simulation times are very long)

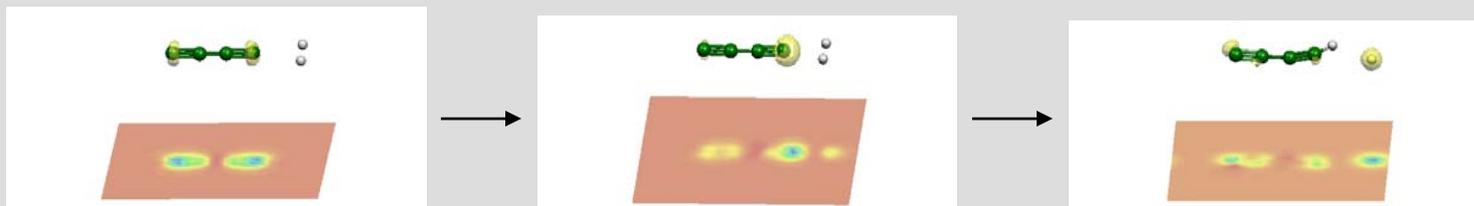


(2) Update atom positions using Newton's laws

$$\bullet R_i(t+\Delta t) \leftarrow 2 \cdot R_i(t) - R_i(t-\Delta t) + \Delta t^2 / (M_i) \cdot F_i(t)$$

# Ab Initio Molecular Dynamics – For Problems Beyond Classical MD Simulations

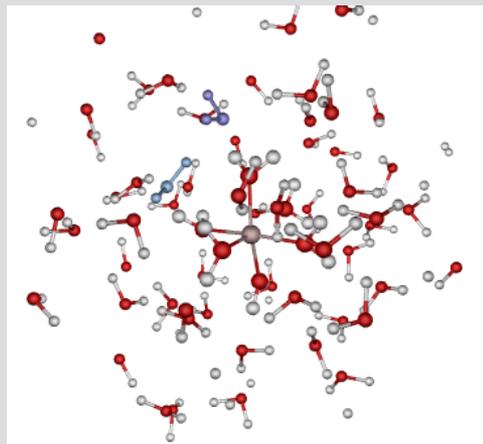
- ▶ Systems with unusual chemical bonding
  - Clusters, surfaces and defects
  - Metallic and semiconductor liquids
  - Diffusion of impurities and defects
- ▶ Phenomena involving changes in electronic structure
  - Band gap of semiconductors in liquid phase
  - Solvation in polar liquids
  - Chemical reactions



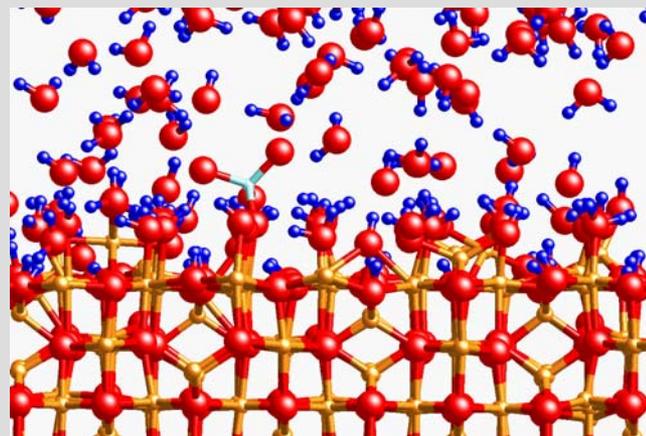
# Application Goals

- Study complex systems with ab initio dynamics

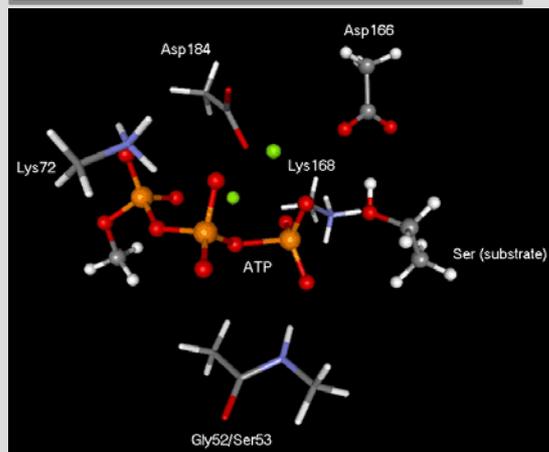
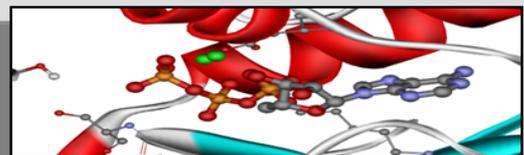
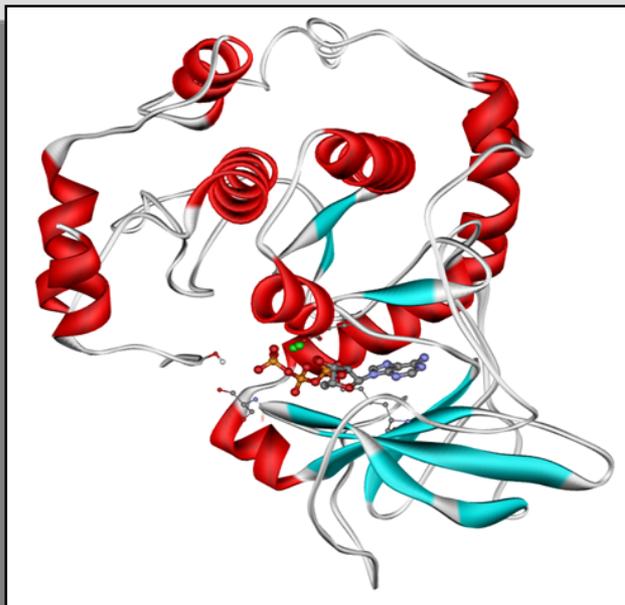
## Al<sup>3+</sup> Hydrolysis



## Active Site on iron-oxide



## Enzymatic Active Site



## Chemical Reactions

# Comparison of Classical and Ab Initio Molecular Dynamics

Classical MD	First Principles MD
✓ <b>Phenomenological potential energy surface</b> (typically restricted to two body contributions)	✓ <b>Potential energy surface calculated directly from the Schrodinger equation</b> (many-body terms included automatically)
✓ <b>Difficult to describe bond breaking/making</b>	✓ <b>Describes bond breaking/making</b>
✓ <b>Electronic properties are <u>not</u> available</b>	✓ <b>Electronic spectra included in calculation</b>
✓ <b>Can do millions of particles</b>	✓ <b>Limited to 250 atoms with significant dynamics</b>

# Basic Features of Ab Initio Molecular Dynamics

Energy functional

$$E[n, \mathbf{R}_I] = \sum_i \left\langle \psi_i \left| -\frac{1}{2} \nabla^2 \right| \psi_i \right\rangle + \int V_{ext}(\mathbf{r}; \mathbf{R}_I) n(\mathbf{r}) d\mathbf{r} + \frac{1}{2} \iint \frac{n(\mathbf{r}) n(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|} d\mathbf{r} d\mathbf{r}' + \int \varepsilon_{xc}(n) n(\mathbf{r}) d\mathbf{r}$$



Ground state energy

$$-\frac{\delta E}{\delta \psi_i^*} = \sum_{\beta} \lambda_{ij} \psi_j$$

Force Calculation



Ion motion

$$M_I \ddot{\mathbf{R}}_I = -\frac{\delta E}{\delta \mathbf{R}_I}$$

# Pitfalls of Ab Initio Molecular Dynamics

▶ Expensive?

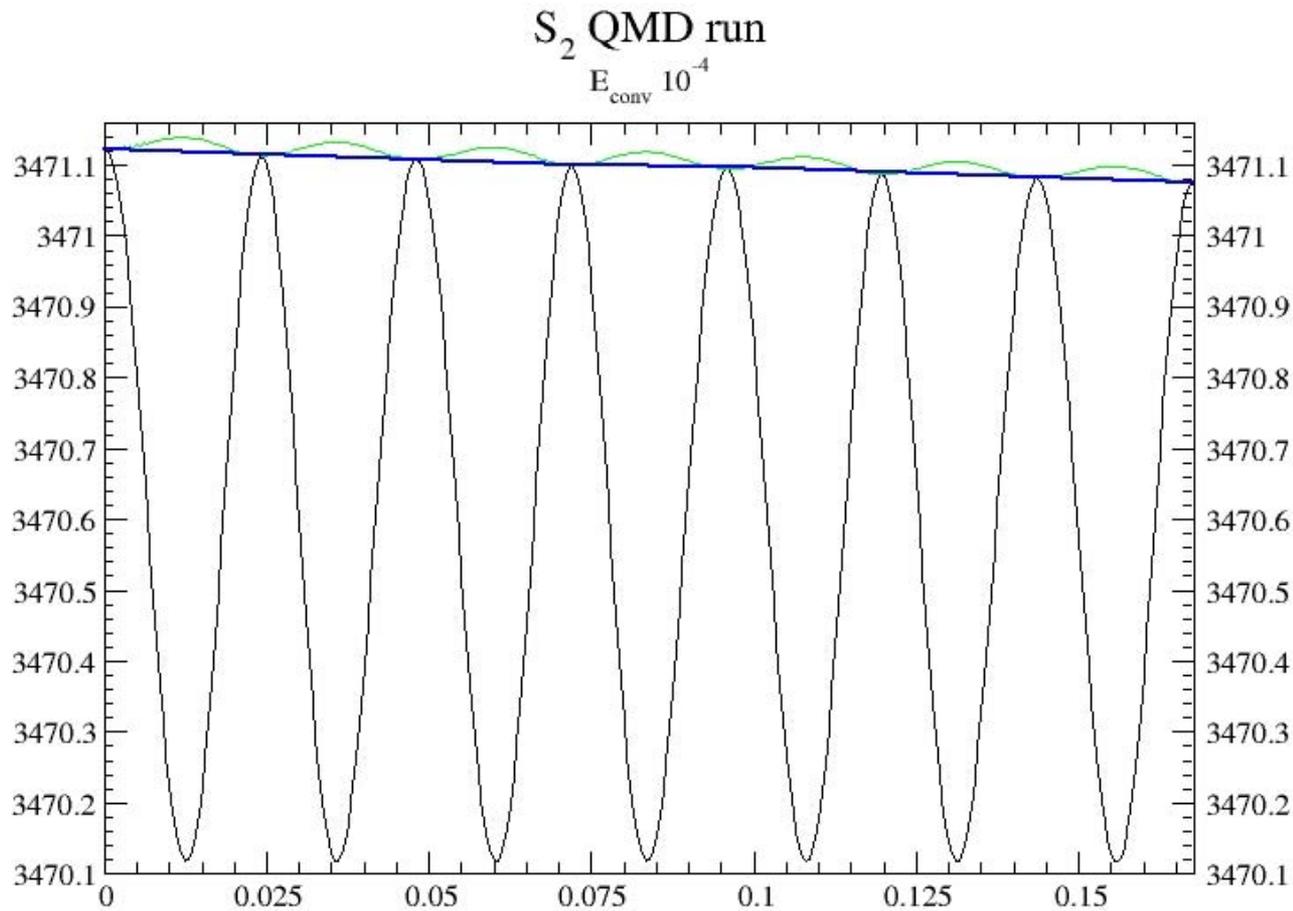
▶ *Energy Conservation – Born-Oppenheimer Error*

$$dE/dR = (\partial E/\partial c)(dc/dR) + \partial E/\partial R$$

*“Attempts to implement such a dynamical scheme in a straightforward fashion prove to be unstable. Specifically, the atomic dynamics do not conserve energy unless a very high degree of convergence in the electronic structure calculation is demanded. If this is not done the electronic system behaves like a heat sink or source.....”*

*-- Remler and Madden*

# $^3\Sigma_g^- S_2$ Energy Surface from QMD Simulation



# Car-Parrinello Dynamics

- ▶ Car and Parrinello suggested that ionic dynamics could be run in parallel with a fictitious electronic dynamics via the following Lagrangean

$$L = \sum_i \frac{1}{2} \mu \langle \dot{\psi}_i | \dot{\psi}_i \rangle + \sum_I \frac{1}{2} M_I \dot{R}_I^2 + E[\{\psi_i\}, \{R_I\}, \text{constraints}]$$

- ▶ Amazingly these equations of motion result in a conservative ionic dynamics that is extremely close to the Born-Oppenheimer surface.
- ▶ The electronic system behaves quasi-adiabatically. That is the electronic system follows the ionic system and there is very little additional motion wandering away from the Born-Oppenheimer surface.

In order to solve the AIMD equations we need to expand the wavefunctions  $\Psi$  in a basis set

$$\Psi_i = \sum_{\alpha} c_{\alpha} \varphi_{\alpha}$$

### Atomic centered basis set (e.g. gaussians, ...)

- All-electron (both core and valence electrons included in calculation)
- Forces are expensive to calculate
- First principles MD expensive

### Plane wave basis set

- Typically requires pseudopotentials (defined on following slide) but can be made all electron with PAW
- Forces can be calculated efficiently
- Efficient first principles MD

# Elements of Ab Initio Molecular Dynamics

## Local Density Functional Theory

$$E = \sum \int \psi_i^*(\mathbf{r}) [-1/2\Delta] \psi_i(\mathbf{r}) d\mathbf{r} + \int V_{\text{ext}}(\mathbf{r}) n(\mathbf{r}) d\mathbf{r} \\ + (1/2) \int \int \frac{n(\mathbf{r}) n(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|} d\mathbf{r} d\mathbf{r}' + E_{\text{xc}}[n, |\nabla n|]$$

$$n(\mathbf{r}) = \sum \psi_i^*(\mathbf{r}) \psi_i(\mathbf{r}) \quad \int \psi_i^*(\mathbf{r}) \psi_j(\mathbf{r}) d\mathbf{r} = \delta_{i,j}$$

# Elements of Ab Initio Molecular Dynamics

## Car-Parrinello Equation - Local Density Functional Theory

if  $\mu=0$  then Kohn-Sham

$$\mu \ddot{\psi}_i = \frac{\delta E}{\delta \psi_i^*} - \sum_j \Lambda_{i,j} \psi_j$$

$$\frac{\delta E}{\delta \psi_i^*} = ((-1/2)\Delta + V_{\text{ext}} + V_c + V_{\text{xc}})\psi_i$$

$$M_I \ddot{\mathbf{R}} = \mathbf{F}_I$$

$$\mathbf{F}_I = \sum_i \langle \psi_i | \frac{\partial H}{\partial \mathbf{R}_I} | \psi_i \rangle$$

**need to be  
efficient**

**We use plane-waves and pseudopotentials**

$$\psi_i(\mathbf{r}) = \sum_{\mathbf{k}} c_{\mathbf{k}}^i \exp(i\mathbf{k} \cdot \mathbf{r}) \quad V_{\text{ext}} \rightarrow \text{pseudopotentials}$$

$$\left[ (-1/2) \Delta \Psi \right] + \left[ V_{\text{ext}} \Psi \right] + \left[ V_{\text{c}} \Psi \right] + \left[ V_{\text{xc}} \Psi \right] = E \Psi$$

$$\left[ \langle \Psi_i | \Psi_j \rangle = \delta_{i,j} \right]$$

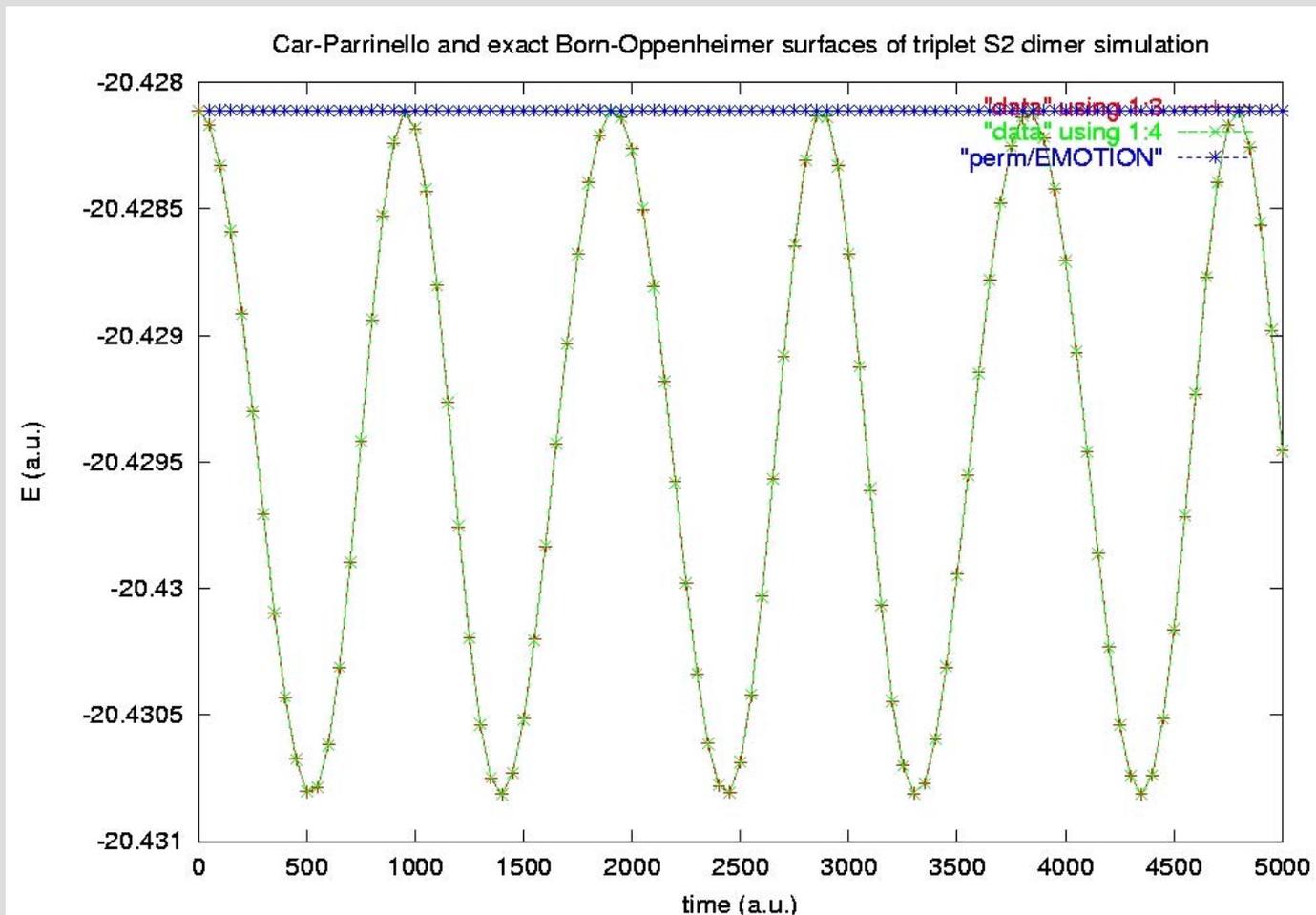
### Scaling for calculating $H\Psi + \Lambda\Psi$

- ▶  $N_e * N_g$  - diagonal in k-space
- ▶  $(N_a * N_g + N_g \text{Log } N_g + N_e * N_g) + (N_a * N_e * N_g)$  - diagonal in r-space and k-space
- ▶  $N_e * N_g \text{Log } N_g + N_e * N_g + 2 N_g \text{Log } N_g + N_g + N_e * N_g$  - diagonal in r-space
- ▶  $N_e * N_g \text{Log } N_g + N_e * N_g + N_e * N_g$  - diagonal in r-space
- ▶ Orthonormalization:  $N_e^2 * N_g + N_e^3$

# Example 1b: S<sub>2</sub> molecule LDA Car-Parrinello Simulation.

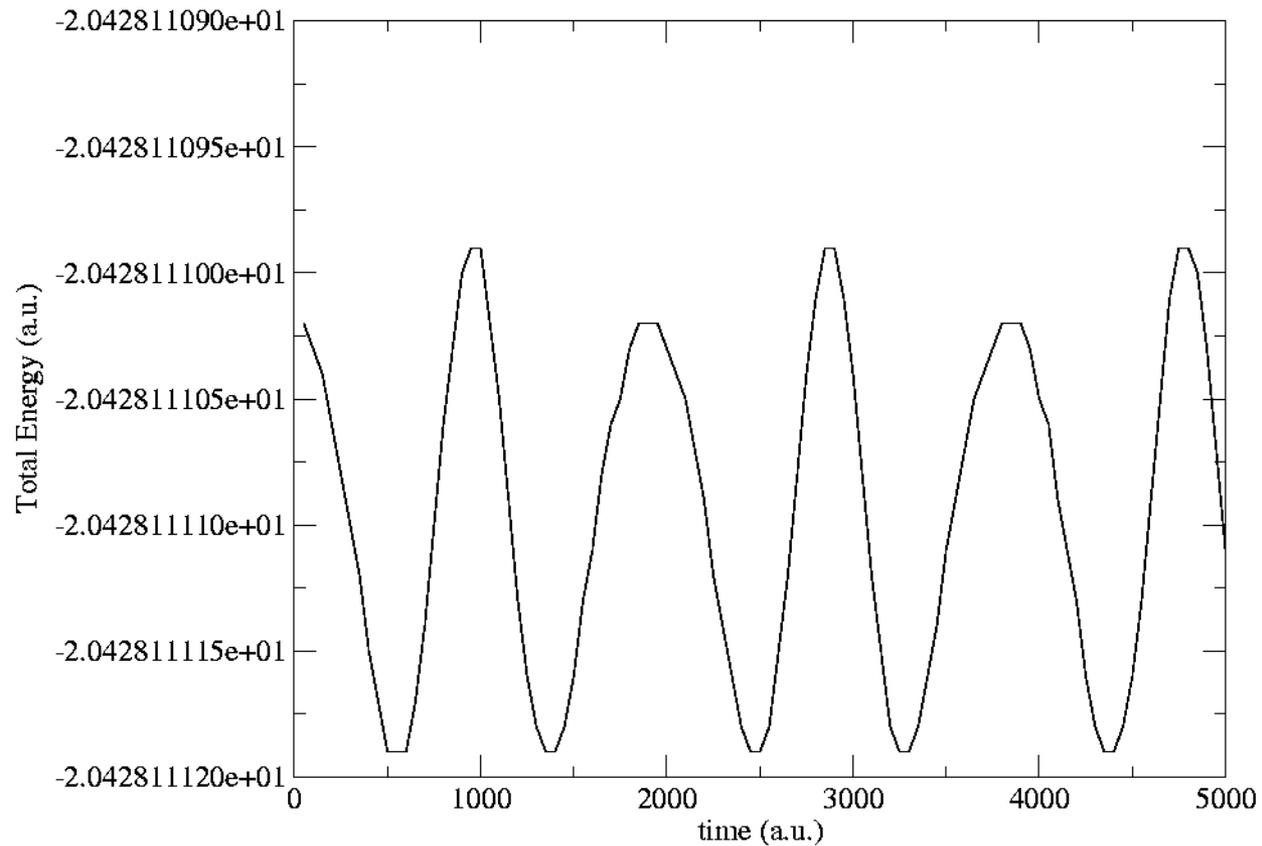
```
title "S2 MD LDA/25Ry"
start s2.md
geometry
S 0.0 0.0 0.0
S 0.0 0.0 1.95
end
pspw
  car-parrinello
    time_step 5.0      #Typically between 1 and 20
    fake_mass 600.0   #Typically between 300 and and 1500
    loop 10 100
  end
  mult 3
end
set nwpw:minimizer 2
task pspw energy
task pspw car-parrinello
```

# $^3\Sigma_g^-$ - $S_2$ Energy Surface from Car-Parrinello Simulation

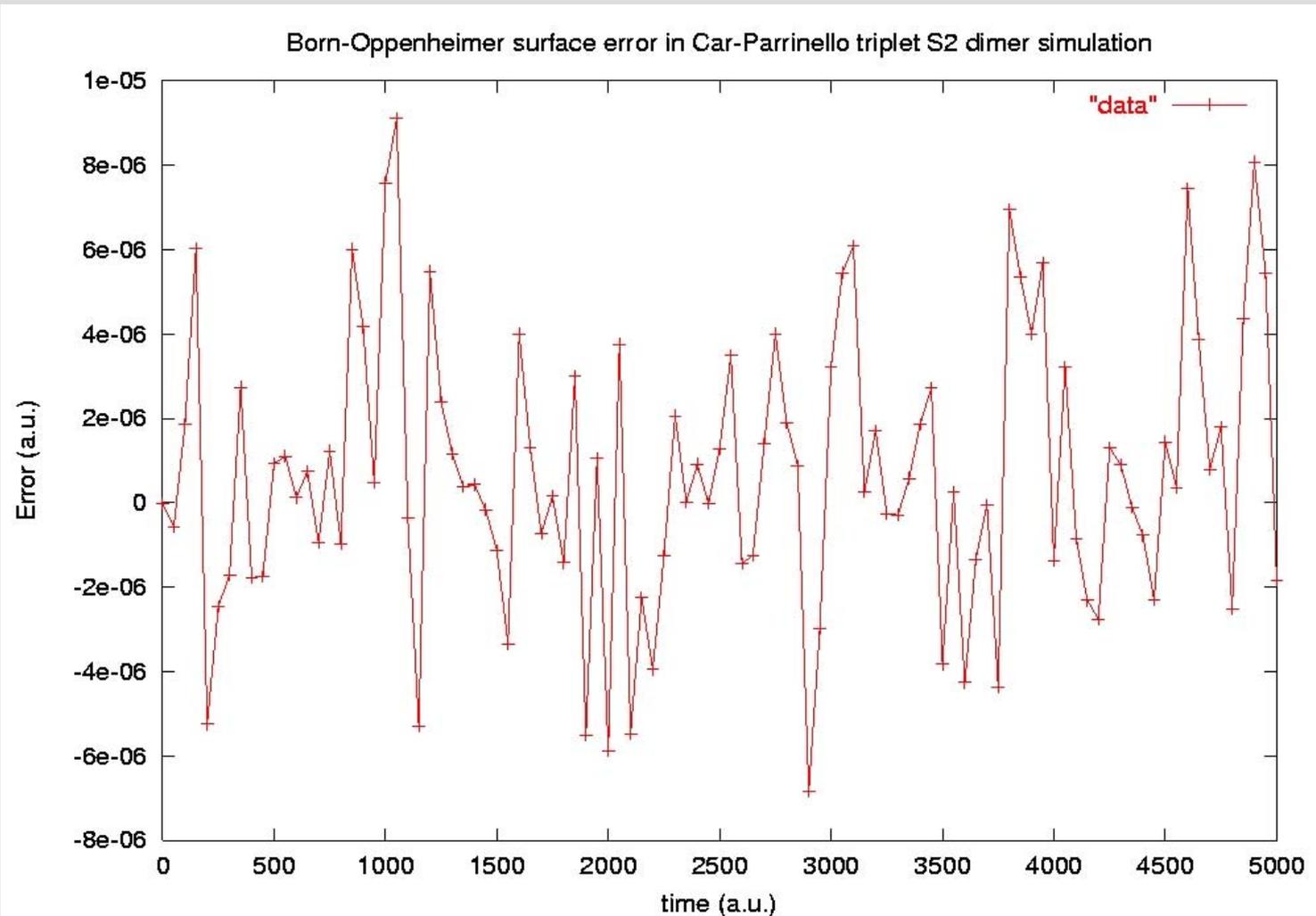


# Energy Conservation

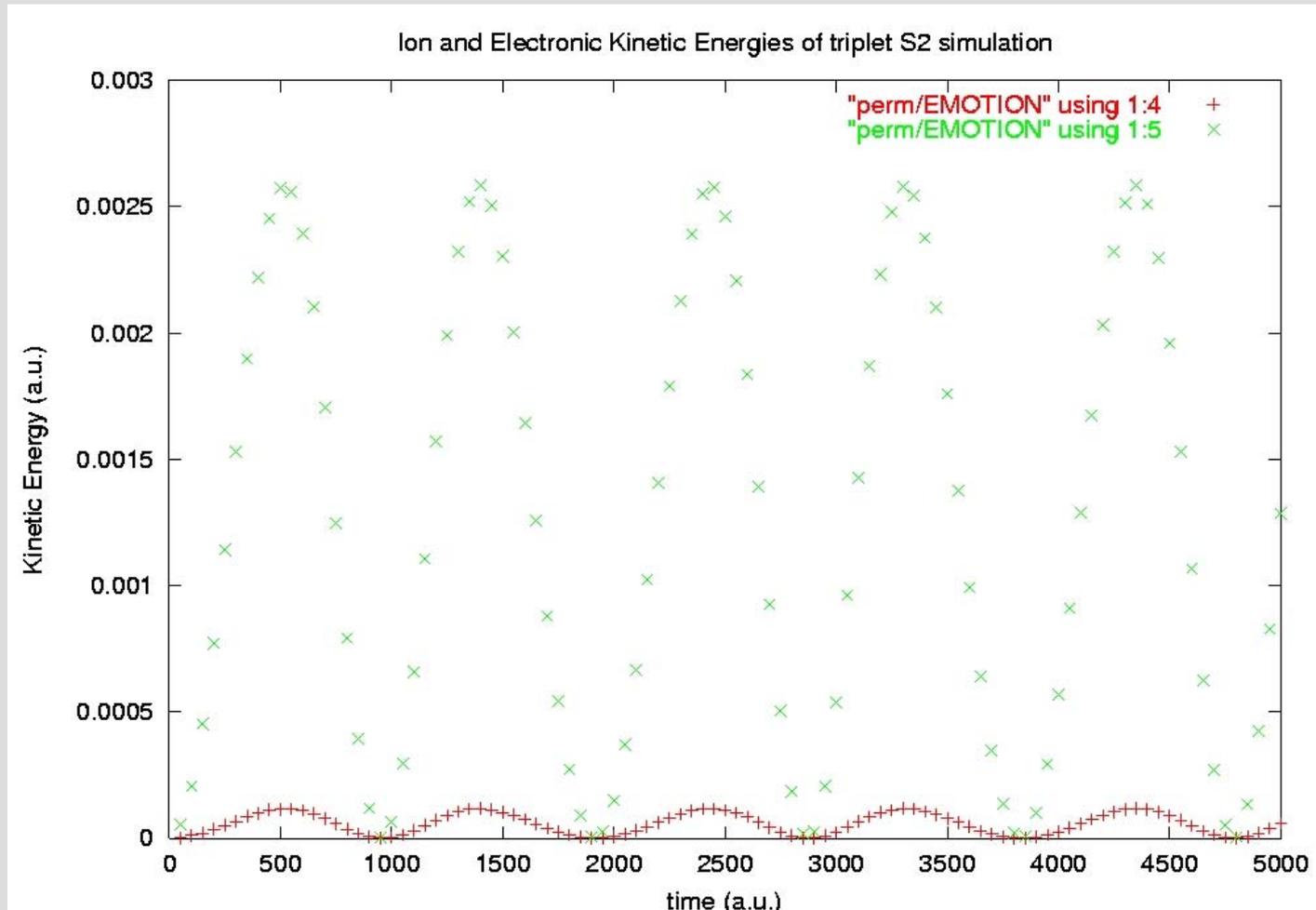
Total Energy Conservation of triplet S2 simulation



# Born-Oppenheimer Error



# Ionic and Fictitious Electronic Kinetic Energies



# A Closer look at Born-Oppenheimer and Car-Parrinello

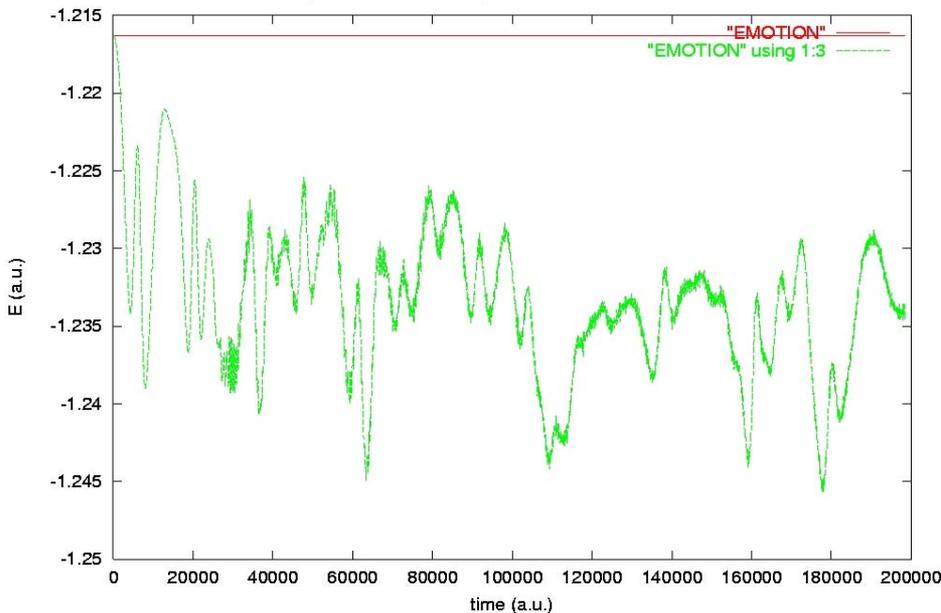
- ▶ Adiabaticity is not built into the Car-Parrinello equations of motion. As pointed out by Remler and Madden

*“equipartition principle tells us that the average kinetic energies of all degrees of freedom in the classical system will be equal at equilibrium. The adiabatic state, in which the fictitious system is at a very low temperature and the ionic system is hot is therefore metastable.”*

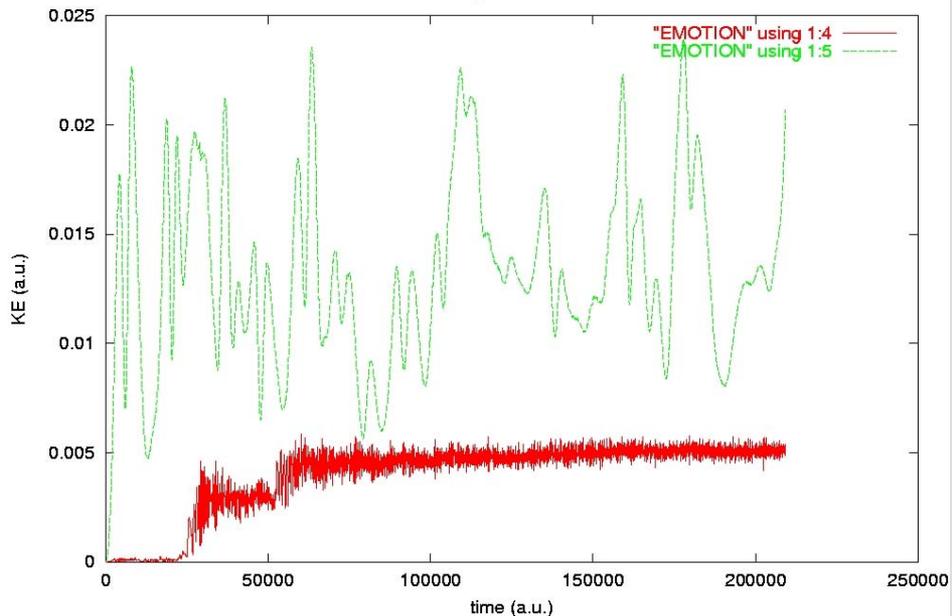
- ▶ The metastable motion is the result of a good start-up procedure and the overlap of the fictitious electronic motion with the ionic motion must be small (i.e. Start simulation on BO surface! Also, standard CP works best for large band gap systems)
- ▶ Total ionic momentum is NOT rigorously conserved

# Na<sub>5</sub> Car-Parrinello Simulation – Electronic heating in Car-Parrinello

Total Energy and Potential Energy from a Na<sub>5</sub> Car-Parrinello Simulation



Electronic and Ionic Kinetic Energies from a Na<sub>5</sub> Car-Parrinello Simulation



Electronic heating can be controlled with thermostats. However, this can result in a serious fluctuation of the total ionic momentum, especially in isolated systems.

# Ab Initio Molecular Dynamics References

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