

Relativity, Spectroscopy and the EMSL Basis Set Library



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Outline



Relativity

- Intro into relativistic effects
- Capabilities in NWChem to handle relativity

Spectroscopy

- NMR properties
- Vibrational frequencies
- EMSL Basis Set Library



Quick introduction to relativity

Dirac Hamiltonian instead of Schrödinger Hamiltonian
 Includes description of positron states







FMS

Effects of relativity



Scalar relativistic

- Contraction and stabilization of s- and p-type orbitals
- Expansion and destabilization of d- and f-type orbitals
- Spin-orbit splitting
 - Orbitals with angular momentum I > 0 split into subshells I $\pm \frac{1}{2}$
 - Coupling between electronic states





Non-relativisticRelativisticElectron density plot of the $7\gamma_{6g}$ spinor in UF₆





Relativity in every day life



- Non-relativistic gold has silver color
 - Stabilization of s-band and destabilization d-band shifts absorption via d-s transition from UV to Vis









Phosphorescence

 Singlet-triplet transitions and surface crossings are allowed due to spin-orbit coupling, i.e. spin is not a good quantum number



 Street lights work with "forbidden" spectroscopic transition ³P₁ to ¹S₀

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NWChem can handle both scalar and spin-orbit effects at the DFT level

task sodft energy

task sodft optimize

task sodft frequencies

- New capabilities under development include
 - Spin-orbit TDDFT for excited states
 - NMR properties (Autschbach, University of Buffalo, USA)



Effective core potentials: scalar



basis

U library crenbl_ecp O library aug-cc-pvdz end

```
ecp
U library crenbl_ecp
end
```

task dft optimize

basis set associated with ECP

effective core potential







еср			
O nelec 2			# ecp replaces 2 electrons on O
O ul	# d		
1	80.0000000	-1.60000000	
1	30.0000000	-0.40000000	
2	1.0953760	-0.06623814	
O s	# s - d		
0	0.9212952	0.39552179	
0	28.6481971	2.51654843	
2	9.3033500	17.04478500	
Ор	#p-d		
2	52.3427019	27.97790770	
2	30.7220233	-16.49630500	
end			





Spin-orbit coupling



In addition to scalar relativistic ECP you need to define a spin-orbit potential

ecp u library stuttgart_rsc_1997_ecp end so						
บ p 2 บ d	9.06055606	14.90142409				
2 u f	8.83183198	2.72712409				
2 end	7.01851629	0.65455772				



All-electron methodologies



NWChem can handle both scalar and spin-orbit effects at the DFT level

relativistic

douglas-kroll on douglas-kroll dkh douglas-kroll dk3full end # Use Douglas-Kroll approximation# Default is Douglas-Kroll-Hess# Third-order Douglas-Kroll

relativistic	
zora on	# zora approximation will be used
end	

Note: You will need all-electron basis sets for ALL elements





Spectroscopy with NWChem



- NWChem has a suite of capabilities for calculating spectroscopic properties
 - NMR properties
 - Electric field gradient (expectation value)
 - Hyperfine coupling (expectation value)
 - Shielding (response property)
 - Spin-spin coupling (response property)
 - Electric polarizability and optical rotation(response property)
- Vibrational frequencies
- UV-Vis already covered earlier



NMR properties



Expectation values						
property efieldgrad hyperfine end	# gets you the electric field gradient tensor # gets you the hyperfine coupling tensor					
Response pro	operties					
property						

shielding	2	12	# calculate shielding tensor for first two atoms
spinspin	1	34	# calculate spin-spin coupling tensor between atoms 3 and 4
end			
task proper	ty		# tell NWChem to run the properties module

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Calculating vibrational frequencies

First optimize your molecule's geometry!!!!

driver	
maxiter 20	# number of geometry optimization steps
tight	# tight convergence for floppy molecules
end	
task dft optimize	# tell NWChem to do the optimization

You can also optimize at the SCF, MP2, TDDFT, and coupled cluster level of theory

- Optimize excited state geometries with TDDFT
- Caution: Coupled cluster optimization will be done using numerical gradients and will be expensive





Masses and intensities

EMSL

NWChem uses the most common masses for elements

task scf frequencies

freq reuse mass hydrogen 2.014101779 mass 3 3.021234 end task scf frequencies

Reuse Hessian# Change the mass for H to D

NWChem only calculates IR intensities

Raman intensities are coming soon



Frequencies output



NWChem prints out two sets of frequencies

- Raw normal modes
- Projected normal modes with translations and rotations projected out
- Use the projected normal modes!

NORMAL MODE EIGENVECTORS IN CARTESIAN COORDINATES						
(Projected Frequencies expressed in cm-1)						
	1	2	3	4	5	6
P.Frequency	0.00	0.00	0.00	0.00	0.00	0.00
1	0.00109	-0.01197	0.12020	-0.07402	0.01164	-0.00799
2	0.00970	0.20350	-0.00123	0.00184	-0.02448	-0.00288
3	0.00624	-0.00875	-0.00516	0.10368	0.00269	0.13042
4	0.00596	-0.00529	0.11985	-0.06716	0.00152	-0.00732
5	-0.03405	0.13917	-0.00045	-0.00078	0.07045	-0.00172





Frequencies output: Eigenvalues



Normal mode frequencies output

Norm	al Eigenvalue	!!	Projected Infra Red Intensities				
Moa			atomic Units	[(debye/angs)**	*2] [(KM/MOI)]	[arbifrary]	
1	0.000		0.000042	0.001	0.041	0.013	
2	0.000	ii	0.003341	0.077	3.257	1.032	
3	0.000	ii	0.000007	0.000	0.007	0.002	
4	0.000	İİ	0.004245	0.098	4.138	1.311	
5	0.000	11	0.002836	0.065	2.764	0.876	
6	0.000	11	0.000063	0.001	0.061	0.019	
7	127.737	ii	0.000163	0.004	0.159	0.050	
8	170.851	ii	0.000049	0.001	0.048	0.015	
9	232.061	İİ	0.000973	0.022	0.948	0.300	

Note: Three rotational and translational modes are zero (projected out)!







Sometimes you get imaginary modes

Norma Mode	l Eigenvalue [cm**-1]	Projected Infra Red Intensities Projected Infra Red Intensities [atomic units] [(debye/angs)**2] [(KM/mol)] [arbitrary]				
			0.009	0 401	0.084	
2	-56.947	0.000814	0.019	0.794	0.171	
3	-34.343	0.004494	0.104	4.381	0.942	
4	-13.396	0.001548	0.036	1.509	0.324	
5	0.000	0.001474	0.034	1.436	0.309	
6	0.000	0.001367	0.032	1.333	0.286	
7	0.000	0.001035	0.024	1.009	0.217	
8	0.000	0.001463	0.034	1.426	0.307	
9	0.000	0.001567	0.036	1.528	0.328	
10	0.000	0.001901	0.044	1.853	0.398	
11	28.105	0.006869	0.158	6.696	1.439	
12	36.721	0.000162	0.004	0.158	0.034	







Causes for imaginary modes:

- Small eigenvalues in floppy molecules may require tight geometry optimization
- If you are searching for a transition state, you should find one imaginary mode
- Side bar: Transition state searches

freq firstneg vardir 4 moddir 1 end task scf saddle

follow first imaginary mode# search along internal variable 4# search along normal mode 1





Frequencies output: Zero-point energy



NWChem prints out zero-point energy and other thermodynamic properties

Temperature	= 298.15K
Zero-Point correctio	on to Energy = 63.909 kcal/mol (0.101845 au)
Thermal correction	to Energy = 67.730 kcal/mol (0.107934 au)
Thermal correction	to Enthalpy = 68.322 kcal/mol (0.108878 au)
Total Entropy	= 75.958 cal/mol-K
- Translational	= 38.765 cal/mol-K (mol. weight = 73.0528)
- Rotational	= 25.463 cal/mol-K (symmetry # = 1)
- Vibrational	= 11.730 cal/mol-K
Cv (constant volum	e heat capacity) = 19.985 cal/mol-K
- Translational	= 2.979 cal/mol-K
- Rotational	= 2.979 cal/mol-K
- Vibrational	= 14.026 cal/mol-K

Note: Different temperature can be set in input





EMSL's Basis Set Exchange





https://bse.pnl.gov/







- Basis Set Exchange is comprehensive online library containing Gaussian basis sets
 - Anyone can download basis sets in the format they want
 - Supporting formats other than NWChem
 - Anyone can contribute basis sets they have developed
 - Only published online after work has been published in literature
- All basis sets that are online are also in the NWChem basis set library
 - Basis Set Exchange is source for NWChem basis set library





Elements covered by basis set



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		"3-21++G" Basis Set Information	
	Summary: VDZD	Valence Double Zeta + Diffuse Functions on All Atoms	
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When you select a basis set

- You see the elements that are covered by the basis set
- You can get more details about the basis set itself



Basis set for certain element



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Summarv:	N/A	"CRENBL ECP" Basis Set Information		
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Select an element

And find the basis sets available for this element





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Click "Get Basis Set"





Adding Basis Sets to the Exchange

- You can add basis sets you have developed to the exchange for download by others
 - Get an account and start adding

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